



Development of a Low-Cost Non-Dispersive Infrared Spectrometer for Real-Time Nitrous Oxide Detection

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Introduction

Why N₂O?

- Nitrous oxide (N₂O) is the third-most problematic greenhouse gas, and is 265 times more potent than CO₂ [1]
- N₂O is the dominant ozone-depleting gas emitted by humans [2]
- Agricultural soils are responsible for 75% of U.S. N₂O emissions [1]
- Monitoring soil N₂O emissions is vital to understanding climate impacts as and promoting sustainable agriculture [4]
- N₂O emissions are highly variable, sometimes changing within hours and across single fields [3, 5]

Project Motivation

- Current N₂O sensing methods are costly and offer limited spacial and temporal resolution [3]
- A small, affordable, robust N₂O sensor can be deployed for real-time data collection to capture nuances in N₂O emissions

Technical Background

NDIR Theory

- N₂O absorbs IR light at 4.5μm
- When N₂O enters the chamber, the intensity of 4.5μm light decreases
- This change is proportional to the concentration of gas in the chamber, given by the Beer-Lambert Law

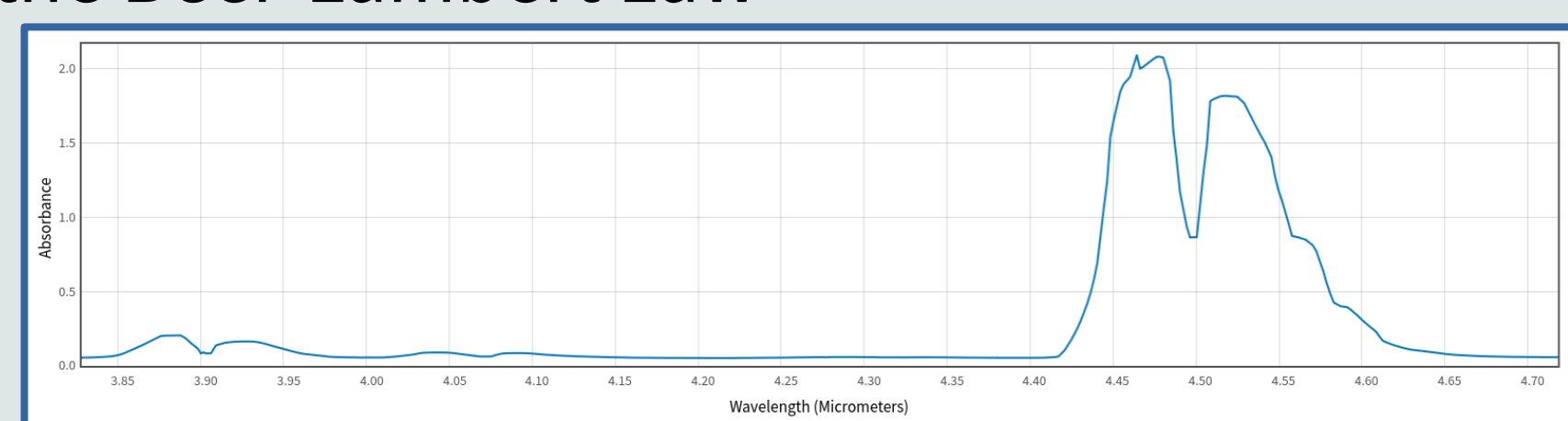


Figure 1: The absorption spectrum of N₂O gas.

The Beer-Lambert Law relates the absorption of light to the concentration of gas, given by:

$$A = \log(I_o / I) = \epsilon lc,$$

A = absorption

I = experimental light intensity, I_o = reference light intensity

ϵ = molar extinction coefficient

l = light path length

c = concentration of target gas

Adjustable Path Length

Since ϵ is constant, $A \sim lc$. When the path length increases, the absorption increases. For low concentrations of gas, increasing the path length improves the sensor's limit of detection.

Sensor Design

- Custom gas cell reduces costs
- Copper pipe used for high IR reflectivity
- Telescoping design allows for adjustable path length for varying N₂O concentrations
- O-ring seals ensure no sample gas escapes
- Modular design allows for easy maintenance and customization

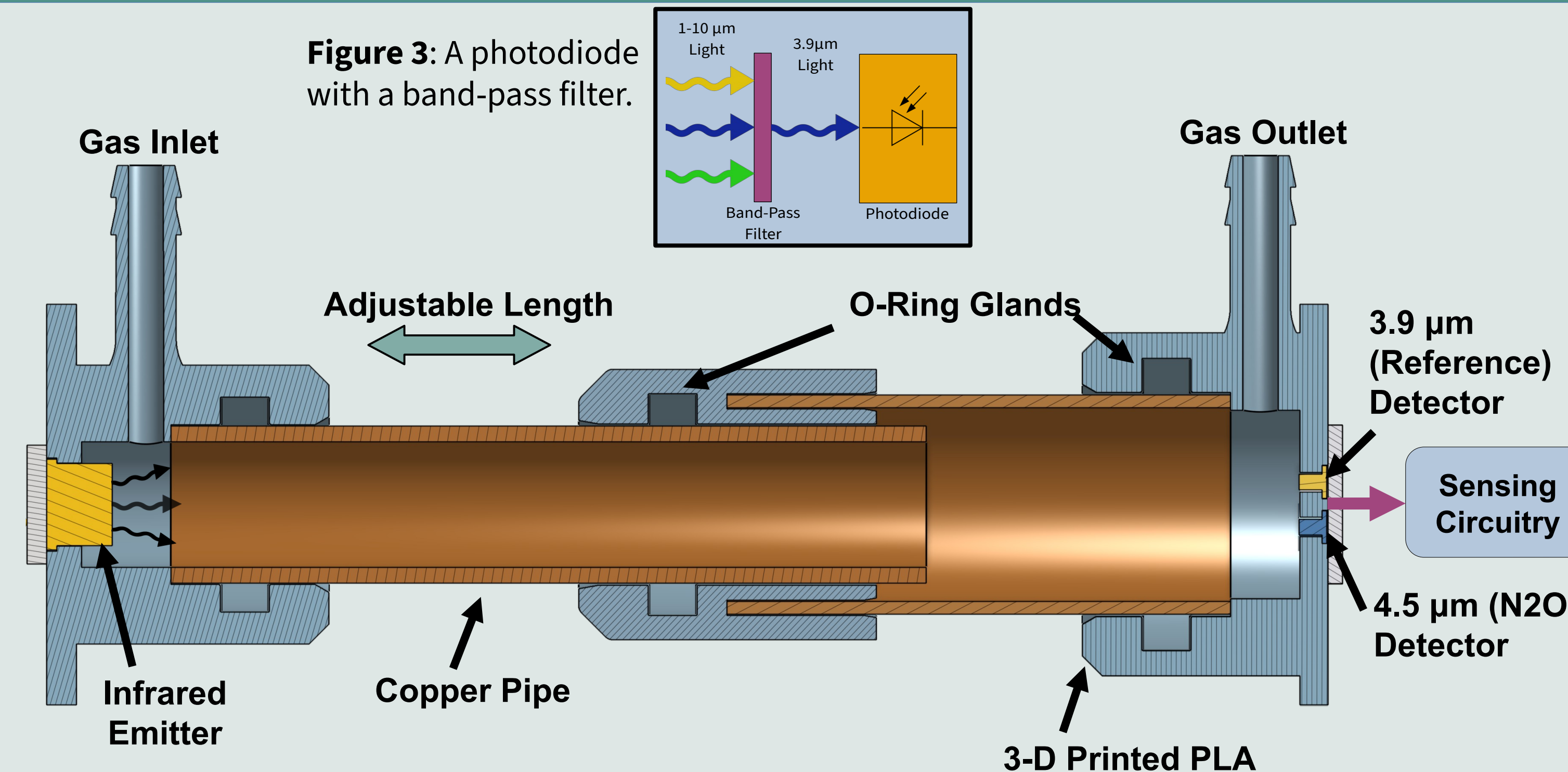


Figure 2: A cross-sectional view of the sensor body CAD model.

Circuit Design

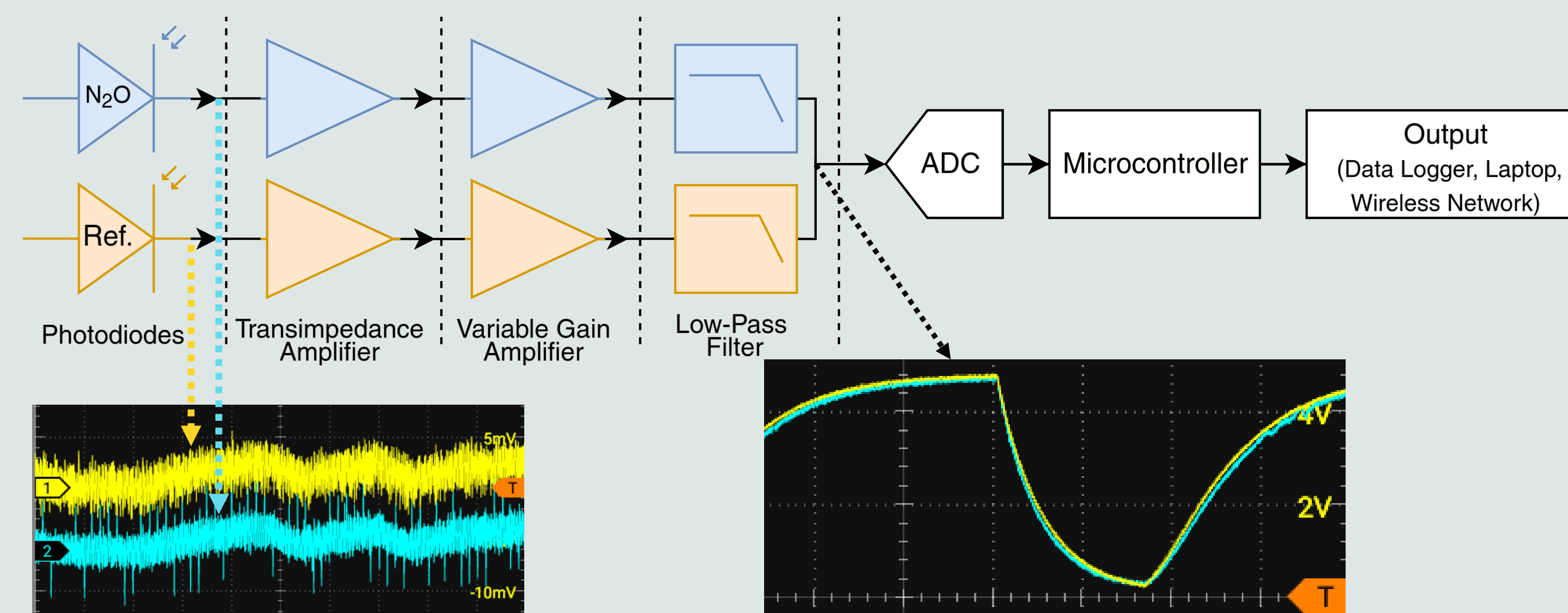


Figure 4: A block diagram representation of the sensor circuitry. A transimpedance amplifier is used to transform the photodiode current into a voltage, and the variable gain amplifier is used to adjust the signal amplitude for maximum ADC resolution.

Preliminary Results

Procedure

- Combined 1 kg healthy soil, 3 tbsp. 21-21-18 fertilizer, and water to saturate soil
- Soil mixture placed on 25°C (77°F) heater
- Gas captured from soil and directed into sensor
- Sensor outlet plugged, so gas accumulates in sensor

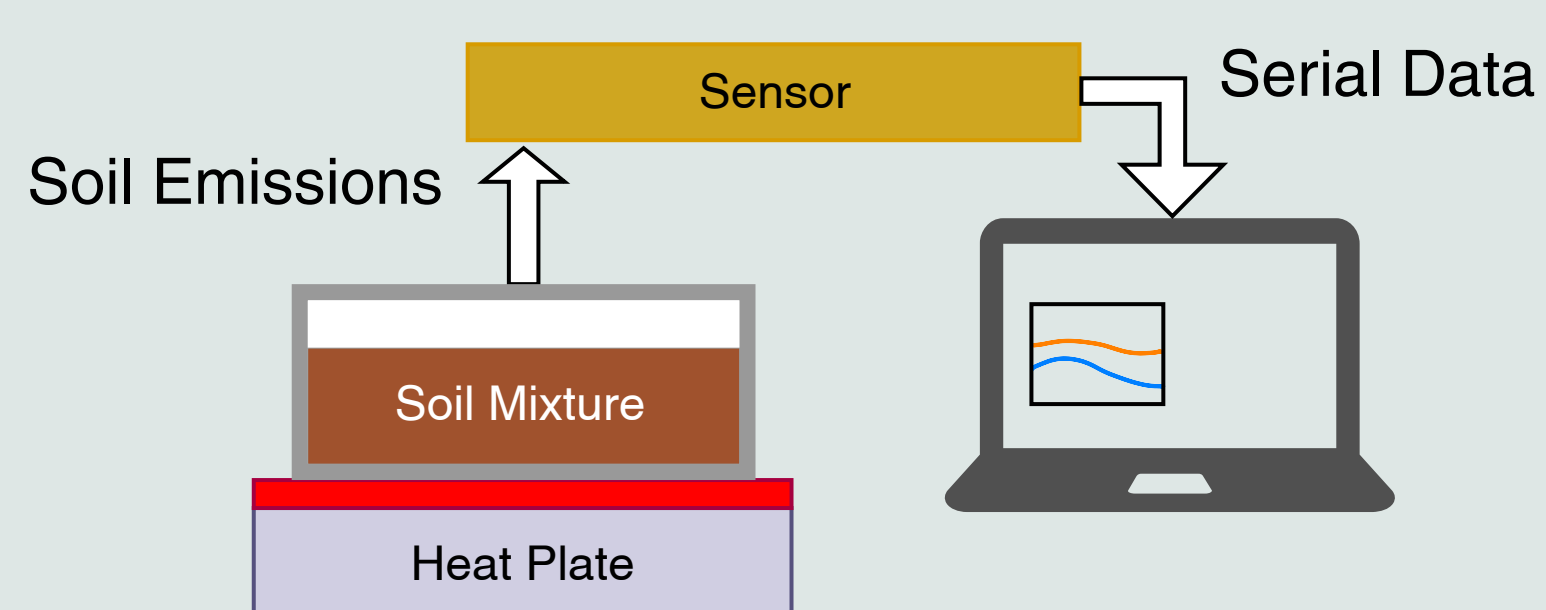


Figure 5: The experimental setup.

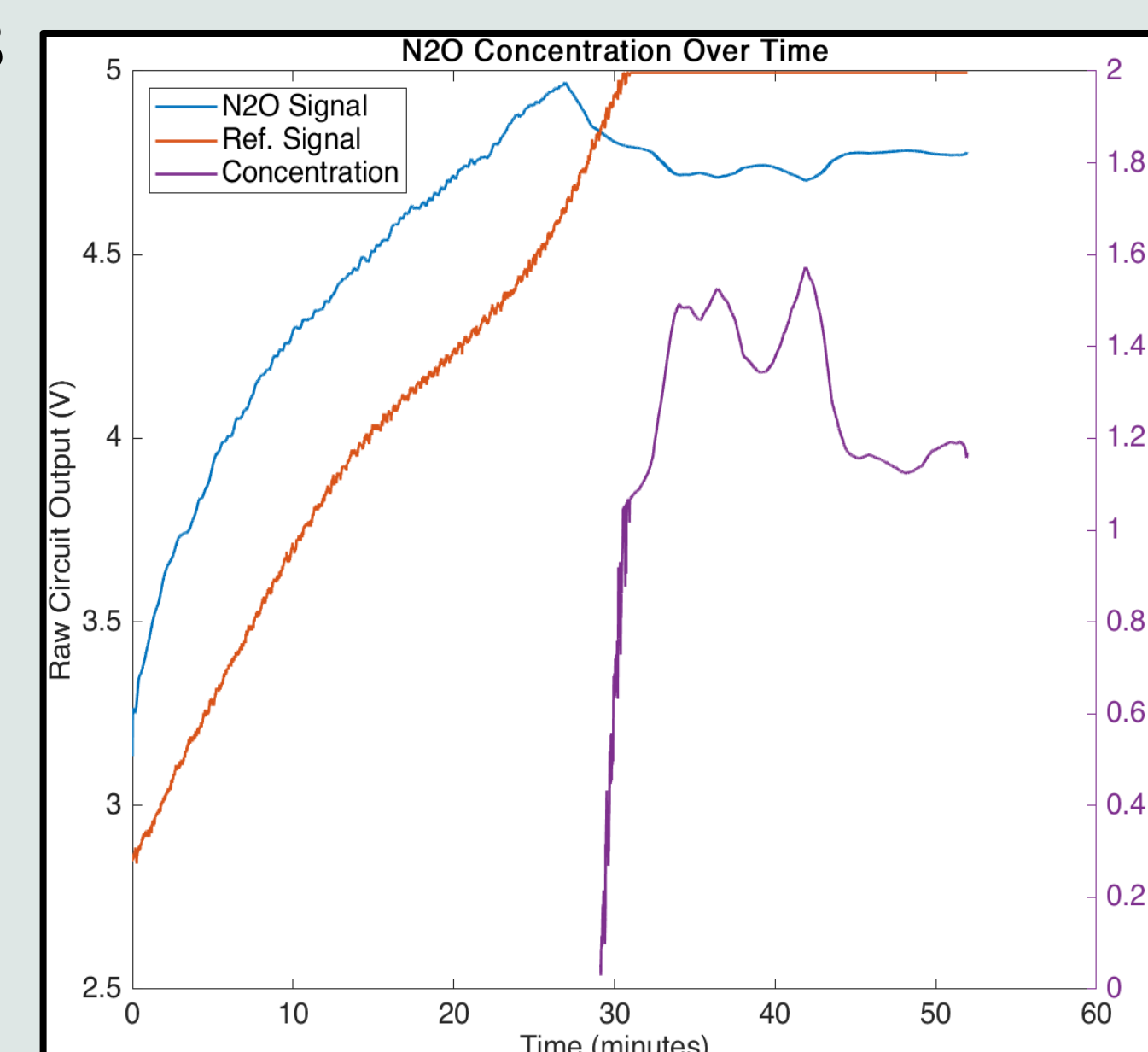


Figure 6: The results of the test.

Discussion

- Shows the process of N₂O detection with this sensor
- Concentration is a rough estimate
- N₂O signal was smoothed, as Arduino sampling introduced noise into the signal
- Generated N₂O soil emissions in a controlled environment
- Further work needed to validate these results

Cost Analysis

Sensor Cost Breakdown	
Part	Cost
Photodiodes	\$228.00
MIR Emitter	\$57.79
Copper Pipe	\$8.05
O-Rings	\$0.30
Microcontroller	\$19.50
Amplifiers and Filters	\$27.20
Other Electronic Parts	\$1.50
Total Cost	\$342.34

Table 1: A breakdown of the sensor costs. Future costs can be reduced using surface mount components.

N ₂ O Sensor Cost Comparison	
Technology	Cost
Quantum Cascade Lasers	\$75,000
LI-COR LI-7820 [6]	\$66,000
Gas Chromatography	\$30,000
Unisense N2O Microsensor	\$6,330
Bandara et al. – “Low-Cost NDIR”	\$2,780
This Device	\$342.34

Table 2: A cost comparison of other N₂O sensing technologies. While each technology has different applications, this design is extremely low cost compared to other designs.

Future Work

- Validate sensor with known N₂O concentrations
- Reduce sensor cost and bulk with SMD circuit parts
- Use LED emitter for increased life and power savings
- Integrate with soil N₂O sampling devices to measure soil N₂O flux
- Combine with a soil sampling device developed in the Real-Time Sensing Lab for total nitrogen measurement
- Add functionality for methane and carbon dioxide sensing

Conclusion

- High spatial and temporal resolution sensors are vital for understanding N₂O emissions
- This design allows for low-cost N₂O sensing that can be adjusted to a variety of applications, lowering the barrier to entry for N₂O monitoring

References

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Acknowledgments

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