



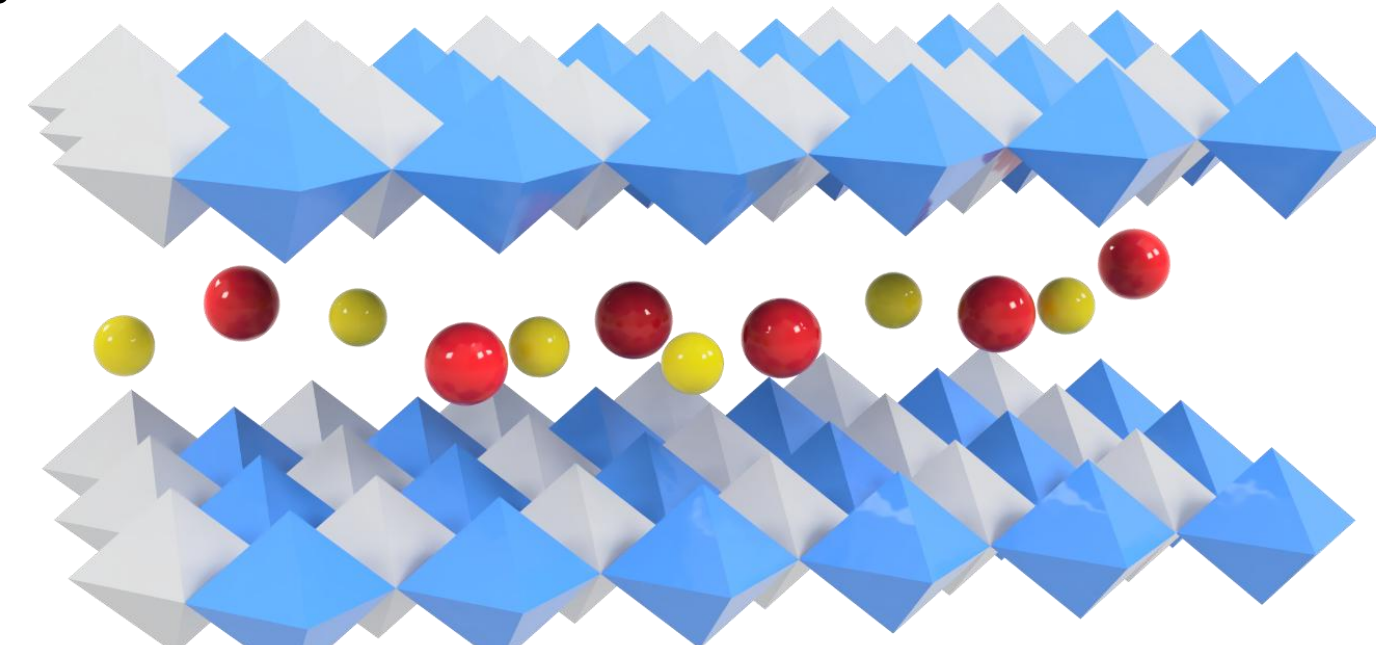
Synthesis and Characterization of Nickel-Titanium Layered Double Hydroxides

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Introduction

- Layered double hydroxides (LDHs): high surface area and unique layered structure



- H_2O
- Interlayer Ions
- $Ti(OH)_4$
- $Ni(OH)_3$

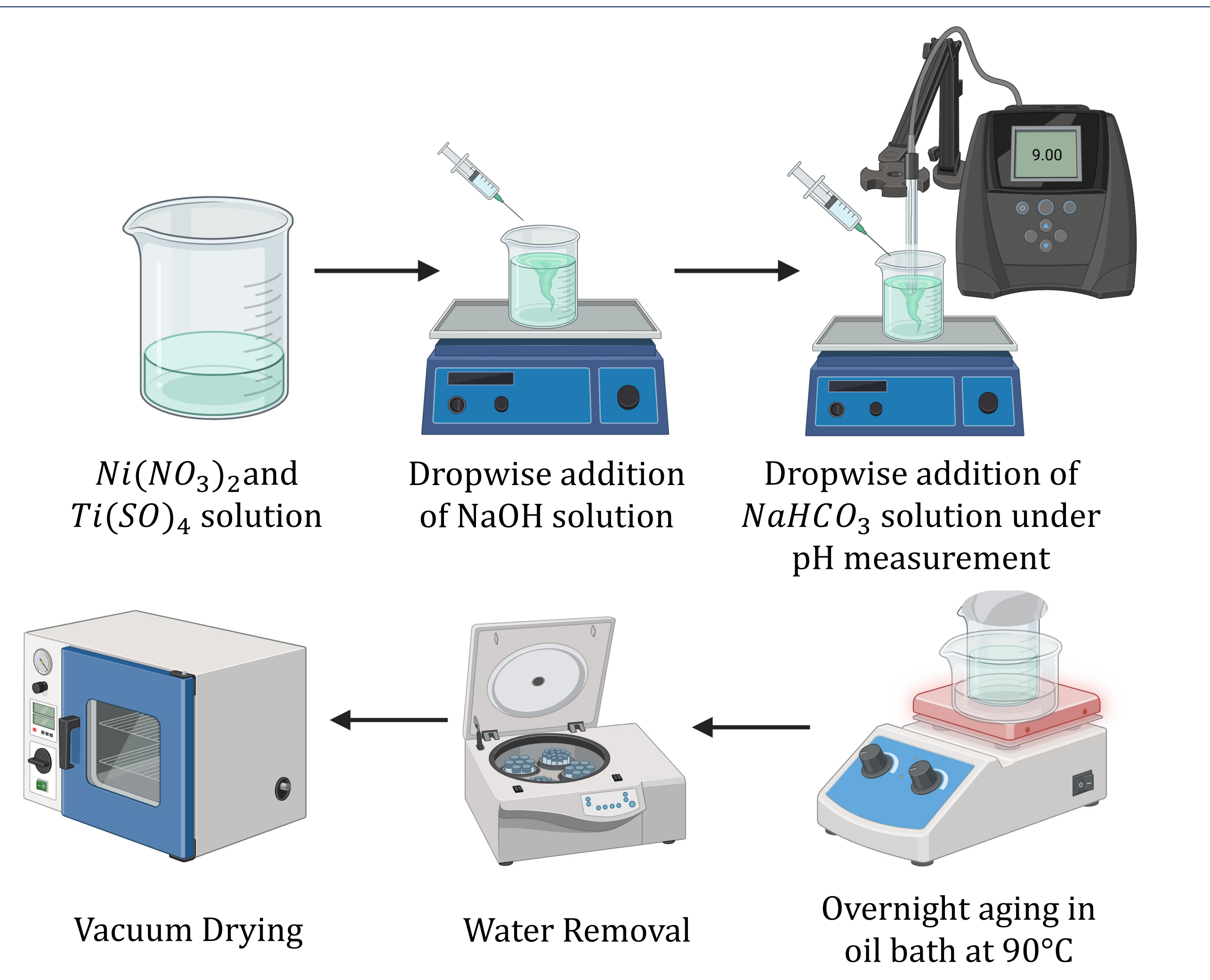
- Nickel-Titanium layered double hydroxides (Ni-Ti LDHs):

various catalytic and environmental applications

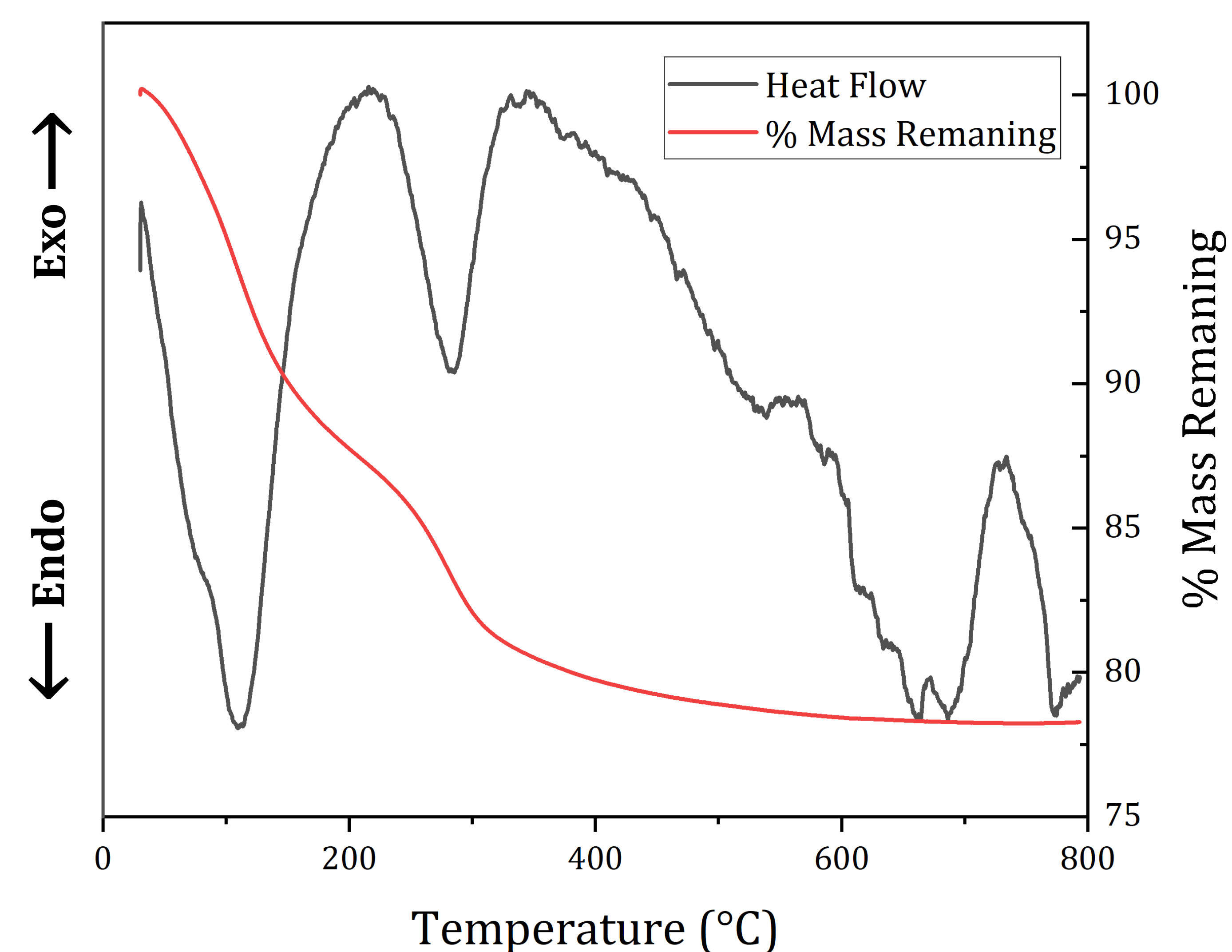
- As-prepared for photocatalysis
- Metal oxides (after calcination) as catalysts for the production of chemicals/fuels

- Research Goal:** Optimize the source of base to control pH and use characterization techniques to understand chemical and physical properties

Catalyst Synthesis

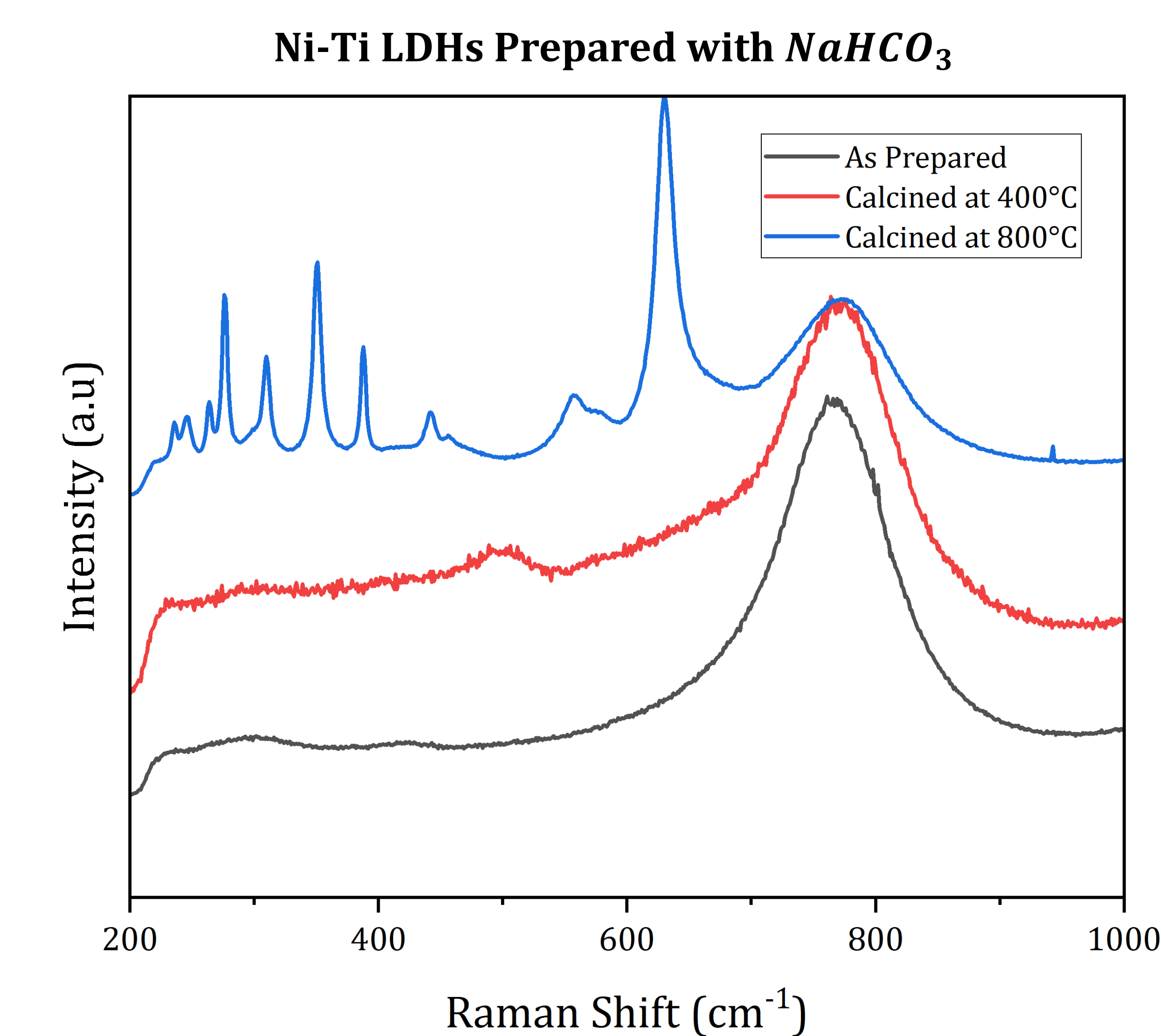


Thermogravimetric Analysis

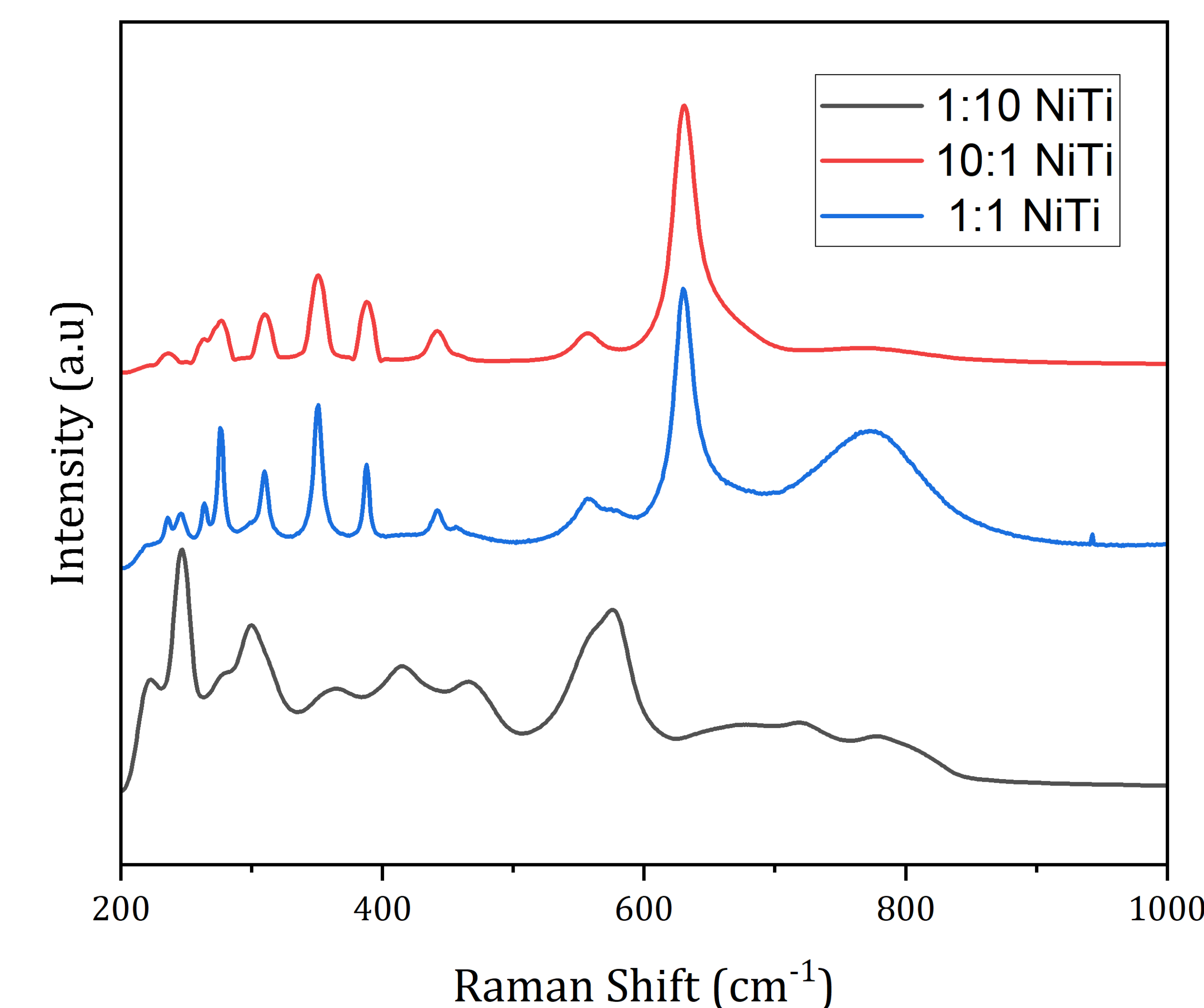


Initial calcination temperature determined to be 400°C

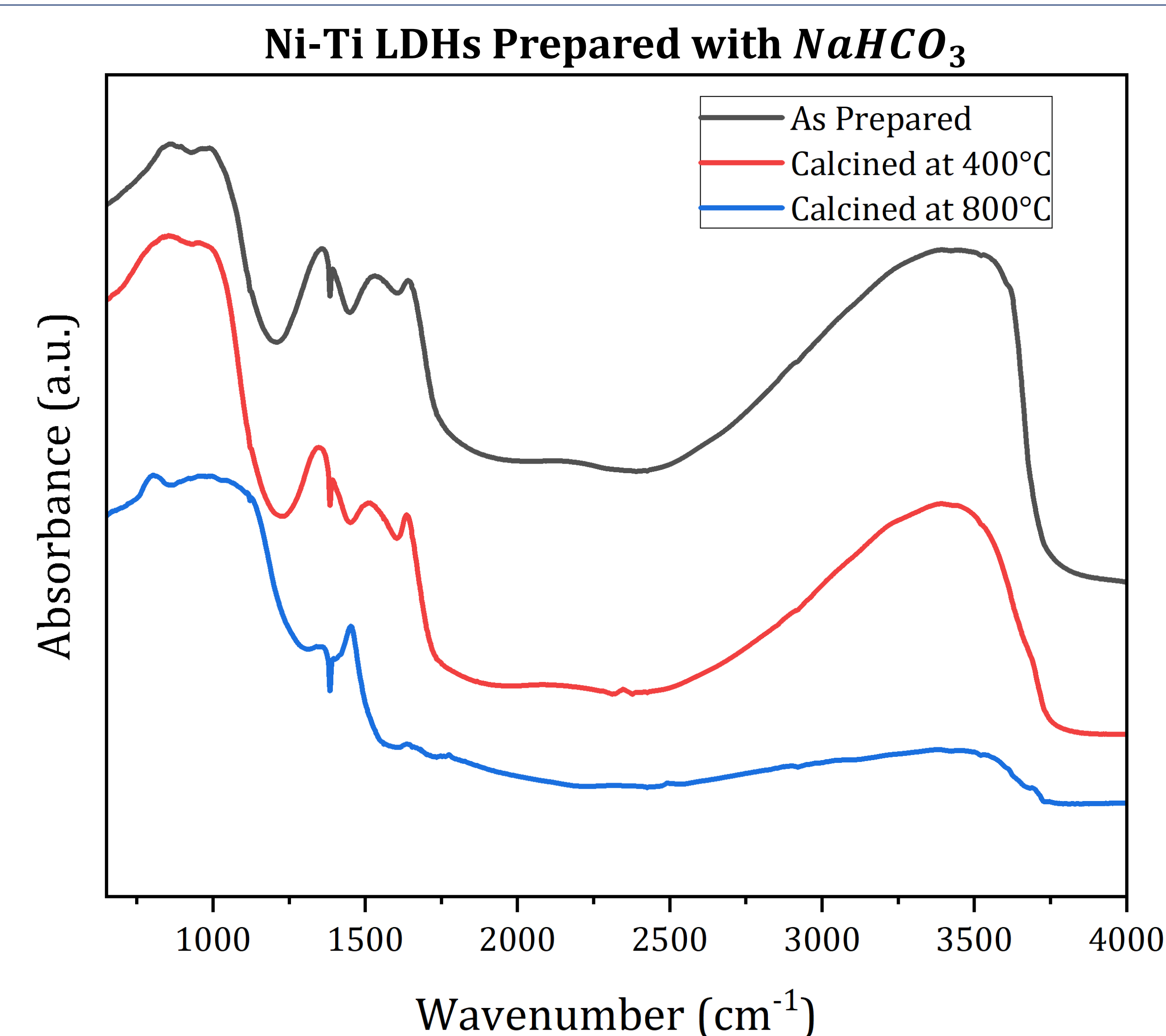
Raman Spectroscopy



Raman Spectra for Various Ratios Calcined at 800°C



Infrared Spectroscopy



Summary

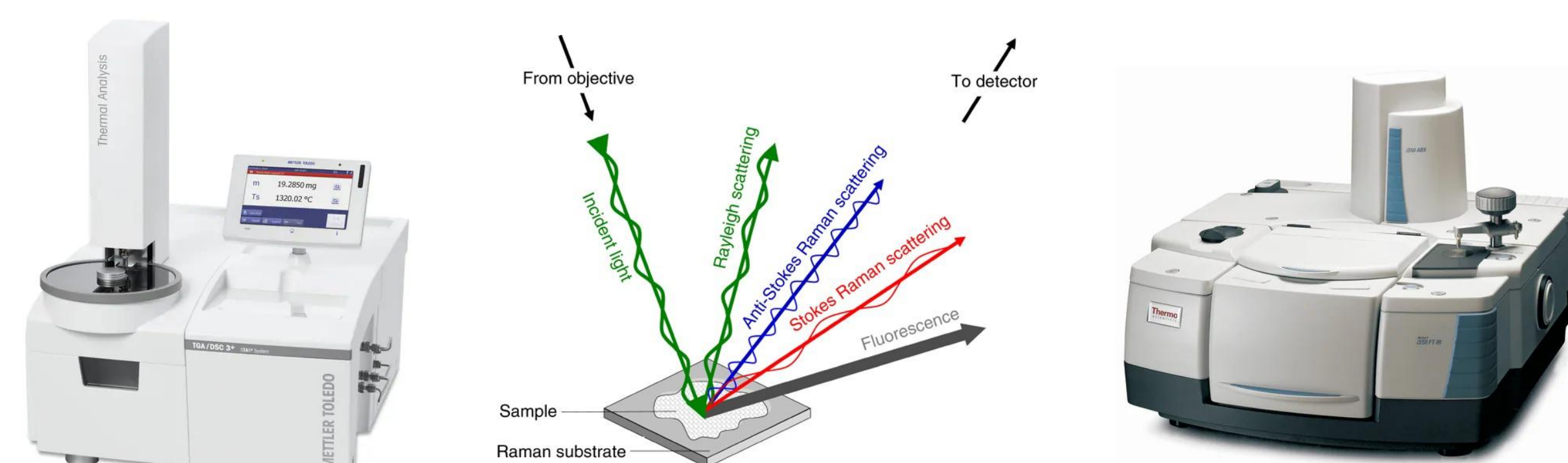
- $NaHCO_3$ is the most optimal base to control pH
- After calcination, the ratio of Ni/Ti determines the structure
- Next Steps:** Application in carbon dioxide conversion

Characterization Techniques

Thermogravimetric Analysis (TGA):
Used to determine an appropriate calcination temperature

Raman Spectroscopy:
Provides insights on chemical structure and crystallinity

Infrared Spectroscopy (IR):
Provides insights on chemical structure and bonds



Acknowledgments

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References

- Layered Double Hydroxides: Present and Future, Vicente Rives, 2001, Nova Science Publishers Inc
- ACS Sustainable Chem. Eng. 2024, 12, 1, 595-609